FODOR, G., AND OTHERS

Determination of the absolute configuration of some tertiary amines and some quaternary ammonium malts. In German.

p. 62. (ACTA UNIVERSITATIS SZECEDIENSIS) Vol. 2, no. 1/4, 1956 Budapest, Hungary

SO: Monthly Index of East European Accessions (FEAI) LC, Vol. 7, No. 3, March 1958

FODOR, G.

A new reaction of the cyclization of amino alcohols; preparation of 20imido-4, 5-cyclopentano-1, 3-oxazolidine. In French.

p. 74. (ACTA UNIVERSITATIS SZEGEDIENSIS) Vol. 2, no. 1/4, 1956 Budapest, Hungary 52 5650

SO: Monthly Index of East European Accessions (EEAI) LC, Vol. 7, No. 3, March 1958

FORDIR, G. HUNGARY/Organic Chemistry - Natural Substances and Their Synthetic Analogues.

G-3

Abs Jour

: Ref Zhur - Khimiya, No 7, 1958, 21601

Author

G. Fodor, I. Sallay, F. Dutka

Inst

Title

: Quaternary Ammonium Salts Derived of (-)-Lupinine.

Orig Pub

: Acta phys. et chem. Szeged, 1956, 2, No 1 -4, 77-79

Abstract

: The configuration of the oxymethyl group with respect to the N atom in (-)-lupinine (I) was studied. Epiner iodides were prepared by the action of CH2ICOOC2H5 (II) on I.

The epimer (III), melting point 154° , $\left[\alpha\right]^{24}D = -49.06^{\circ}$ (c = 1.591), was prepared at about 20° of 1.07 g of I and 1.284 g of II in 3 mlit of absolute C6H6. The epimer (IV), melting point 148 to 150°,

1.05), was obtained of 0.354 g of I and 0.428 g of II in

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Univ Szeged, Hung.

HUNGARY/Organic Chemistry - Natural Substances and Their Synthetic Analogues.

G-3

Abs Jour

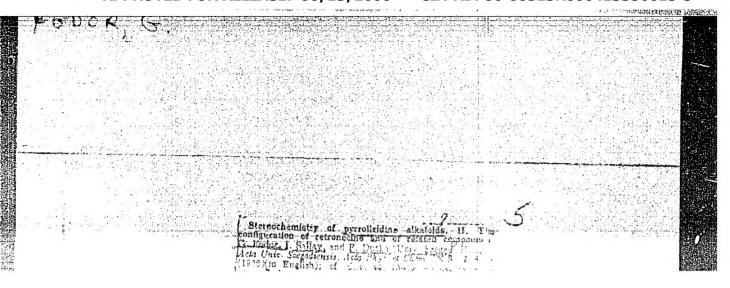
: Ref Zhur - Khimiya, No 7, 1958, 21601

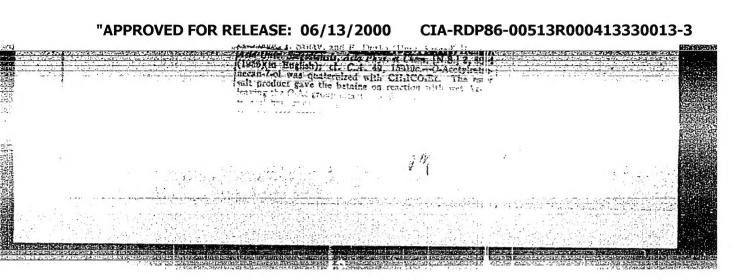
1 mlit of absolute alcohol and 3 mlit of absolute CoHo at 950 (22 hours in a sealed tube, after which 43 hours at about 200). III was transformed into a substance (V), mel-

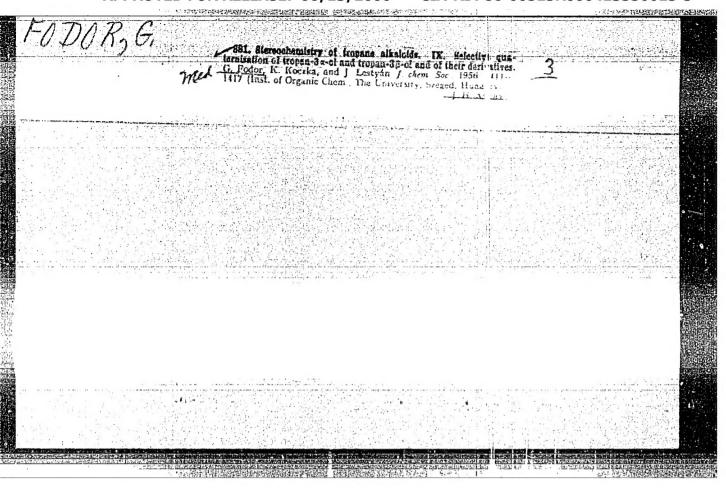
ting point 150 to 152°. $\triangle J^{24}D = +16.67^{\circ}$ (c = 1.5), by boiling in 10 mlit of water in a sealed tube (24 hours at 95°). III produces betaine, melting point 244°,

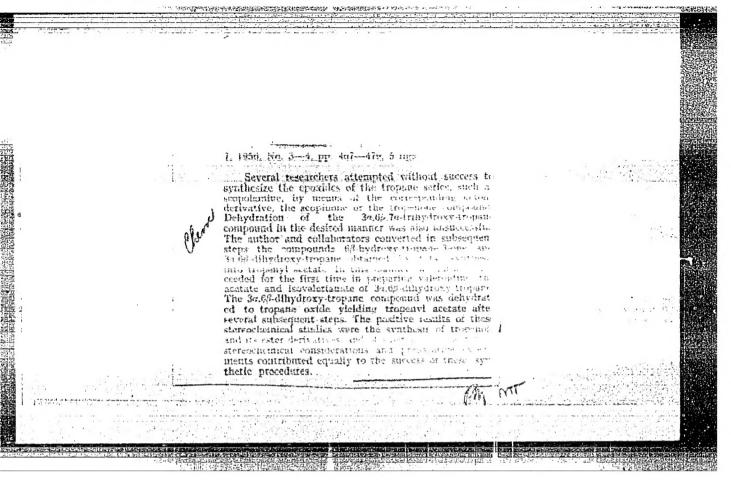
 $\int d^{26}D = +9.91^{\circ}$ (c = 1.029), by the action of an excessive amount of Ag_2O in 25 mlit of water (2 hours of shaking) and following boiling (5 hours). The boiling of betaines derived of III, IV and V with HBr or HI does not result in lactonization. All $\int d^{2}D$ -s were measured in water.

Card 2/2









FODOK, 1.

USSR / Organic Chemistry. Theoretical and General Problems of Organic Chemistry.

E-I

Abs Jour : Ref Zhur - Khimiya, No 6, 1957, No 18994

Author

Fodor G., Kochka K., Leshtian I., Tot I., Khal'mosh G.,

Kovach 'O., Vinche V.

Inst

1 Not given

Title

: *bsolute Configuration of Some Tertiary Amines and Tetra-

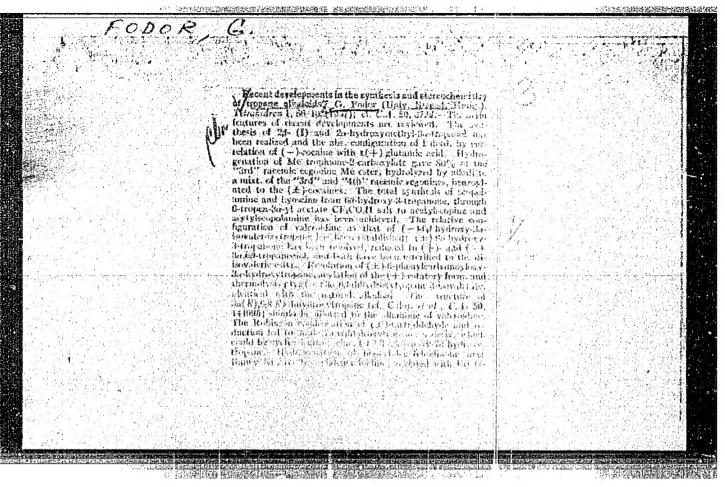
amonium Salts.

Orig Pub : Uspekhi khimiyi, 1956, 25, No 7, 894-902

Abstract : Review of the work by the authors on the study of the spherical orientation of the bonds of nitrogen and the determination of absolute and relative configuration of tertiary amines and salts of tetraamonium bases in Bitliography with 24 titles.

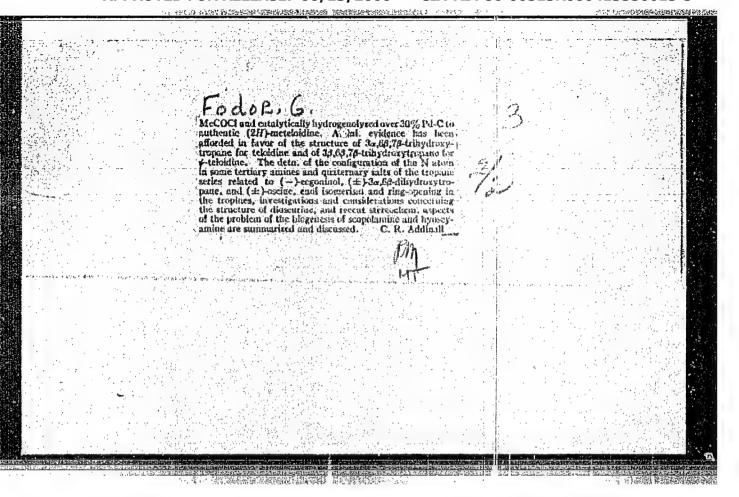
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CIA-RDP86-00513R000413330013-3



HUNGARY/Organic Chemistry. Theoretical and General Questions on Organic Chemistry.

G-l

Abs Jour: Ref Zhur-Khim., No 13, 1958, 43177.

Author : Fodor Gabor, Kovacs Odon, Toth Jozsef, Koczka Karoly, Koczor Istvan, Vincze Iren W., Lestyan

Janos, Halmos Miklos, Dobo Pal.

Inst

: Recent Methods and Advances in Stereochemistry of Title

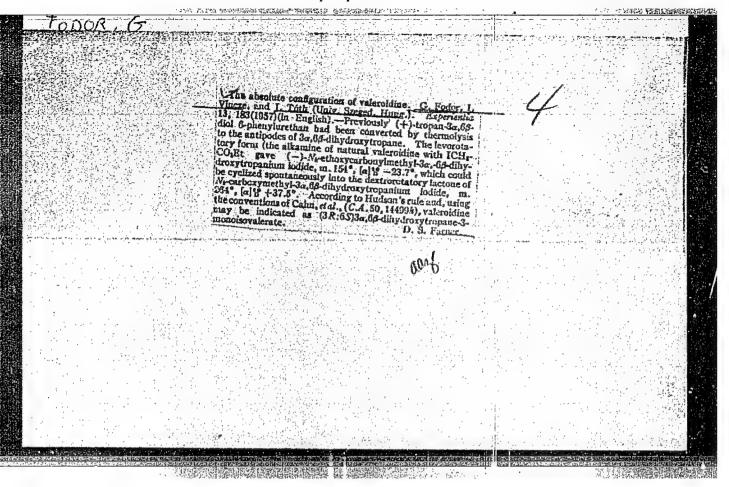
Organic Compounds.

Orig Pub: Magyar tud. akad. Ken. tud. oszt. kozl., 1957, 9,

No 1, 77-91.

Abstract: A review, mostly of the work of the athors. Dibliography 58 references.

Card : 1/1



ACTA CHINTCA

Academiae Scientiarum Hungerinze

Vol. 13, Nrs. 1-2, 1957

THE STEREOCHEMICAL COURSE OF THE CONVERSION

OF 2-UREIDO ALCOHOLS INTO ONAZI LIDINES. II

REARRANGEMENT OF N-THIOUREIDO / LCOHOLS

K, KOCZKA and G, FODOR

(Institute of Organic Chemistry, University of Surged)

Received Auril. 2, 1956

SUMMARY

The aterpechemical course of the conversion of N-thiocarpoin 1-2 units alcohols perticularly that of Orbentovin-hibosoft-mayleibelthic lines alcohols perticularly that of Orbentovin-hibosoft-mayleibelthic lines and a thiocarpoin of the thiasphoton with y-sphetrics) the sparing between 1 at the configuration of the thiasphoton with y-sphetrics) the sparing between 1 at the configuration of the thiasphoton with y-sphetrics) the sparing between 1 at the configuration of the thiasphoton with y-sphetrics) the sparing between 1 at the configuration of the thiasphoton with y-sphetrics) the sparing between 1 at the configuration of the thiasphoton with y-sphetrics and a thiocarpoint of the indical persectively.

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	Vol 13, Wrs 1-2, 195	
	SYNTHETIC CONFIRMATION OF T	II MPOGLERIEGE
A CANADA	OF N=O ACYL MIGRA	fors
	PREPARATION AND REARRANGEMENT OF	
	2.5-DIPHENYL-3.4-DIMETHYL-µ-HYDROXY-1.3 0	MAZOLIDINE
	K. Kocces and G. Foneit	
	Unstitute of Organic Chemistry, University of	L Steedt
	Received April 3, 1956	37.4
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YUGOSLAVIA/Organic Chemistry. deparal and Theoretical Topics of Organic Chemistry.

Abs Jour: Ref Zhur-Khimiye, No 22, 1958, 73924.

: Gabor Fodor, Eva Fodor-Vraga, Arpad Furka. Author

: A Kinetic Contribution to the Knowledge of Title Carbon Rings.

Inst

Orig Pub: Croat. chem. acta, 1957, 29, No 3-4, 303-312.

Abstract: With a view to investigate the influence of spatial factors on the mechanism of N-0 transposition of the acyl group in N-substituted Ov -amine alcohols, the rearrangement of cis- and trans-2-benzamidocyclohexanols-1 (I and II) and cis-2-benzamidocyclopentanol-1 (III) into ois- and trans-2-benzoyloxycyclohexylam amines and cis-2-benzoyloxycyclopenylamine corres-

YUGOSLAVIA/Organic Chemistry. General and Theoretical Topics of Organic Chemistry.

Abs Jour: Ref Zhur-Khimiya, No 22, 1958, 73924.

pondingly under the action of HCl in dioxane was studied. The rate of the reaction with III was measured at 12 to 42° by the determination of the free amine, and that of the reactions with I and II were measured at 71 to 91° by the alkalimetric titration of the excess of HCl as well. Comparing the data for I, II and III after extrapolating them to 25° with the bibliographical values of the reaction rates of N-benzoyephedrine, cis- and trans-2-acetamidocyclohexanols-1 and cis- and trans-2-N-acetylinozamins (IV), the authors arrive at the conclusion that the transposition rate is determined mainly by the structure of the carbon framework of the alcohol, but not by the character

Card : 2/5

YUGOSLAVIA/Organic Chemistry. General and Theoretical Topics of Organic Chemistry:

Abs Jour: Ref Zhur-Khimiya, No 22, 1958, 73924.

of the solvent or of the migrating group, the rate ratio of the arylalifatic, cyclopentanic, cyclo-hexanic and isoaminic derivatives being 1000: 1000: 20: 1 correspondingly. The lesser reaction rates of I and II as compared with III is explained in accordance with the magnitudes of thermodynamic potential changes (I = 24.0, II = 24.3, III = 20.2 kcal per mole) by a lesser probability of intramolecular collisions in the cases of I and II in consequence of the existing conformation equilibrium. The cis-forms are 4 to 6 times more reaction capable than the trans-forms, because the latter can regroup only at the di-E arrangement of the amino and oxy groups, while the E,A, as well as the A,E conformations react in the cis-forms.

Card : 3/5

YUGOSLAVIA/Organic Chemistry. General and Theoretical Topics of Organic Chemistry.

Abs Jour: Ref Zhur-Khimiya, No 22, 1958, 73924.

The above is confirmed by the difference between the activation energy values \triangle E (I - 15.02, II - 17.21 kcal per mole). The value of \triangle E of III (12.89 kcal) corresponds seemingly only to the transposition energy of the aci-group, and the increase of \triangle E of I and II is caused by the energy of the conformation conversion. The proposed mechanism of the regroupment with configuration preservation consists in an electrophilic attack by the proton of the carbonyl 0 and a following nucleophilic attack by the hydroxyl 0 of the carbonyl C with the formation of an intermediary cyclic complex. In accordance with the above, the little reaction capacity of IV can be explained by the difficulty of a nucleophilic attack

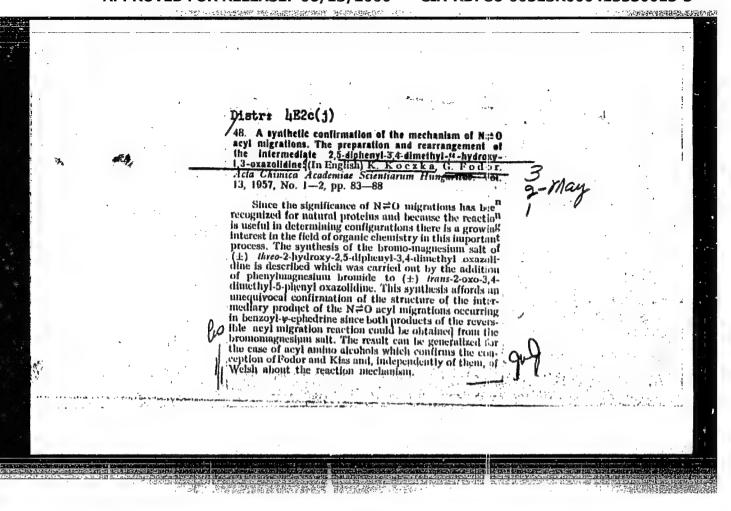
Card : 4/5

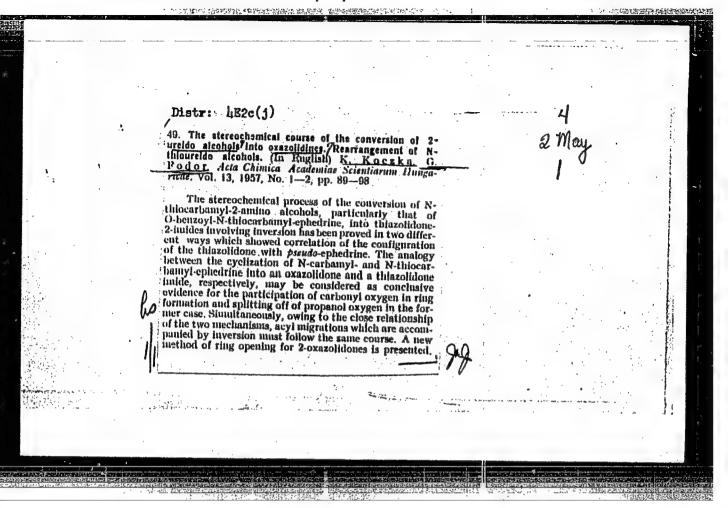
YUGOSLAVIA/Organic Chemistry. General and Theoretical Topics of Organic Chemistry.

Abs Jour: Ref Zhur-Khimiya, No 22, 1958, 73924.

in consequence of the participation of the hydroxyl 0 in the formation of the hydrogen bond. The reaction of the corresponding III trans- derivative does not agree with that mechanism and could not be studied, because it proceeds with a Walden inversion.

Card : 5/5





HUNGARY/Analytical Chemistry. Analysis of Inorganic

Compounds.

Abs Jour: Ref Zhur-Khimiya, No 21, 1958, 70551.

Author Fodor

Inst

: Akad. Kem.

Title

: Determination of Uranium by a Combined Method of

Ion Exchange and Complexometry.

Orig Pub: Magyar tud. akad. Kem. tud. Oszt. Kozl., 1958,

9, No 4, 463-470.

Abstract: No abstract.

: 1/1 Card

HUNGARY/Chemical Technology. Chemical Products and Their Applications. Industrial Organic Synthesis.

Н

Abs Jour: Ref Zhur-Khim., No 8, 1959, 28446.

Author : Fodor, G. and Beregi, L., and Kalley, F.

Inst : Hungarian Academy of Sciences.

Title : Results from Investigations on the Chemistry of

Furan in Hungary.

Orig Pub: Acta Chim Acad Sci Hung, 15, No 3, 315-323 (1958)

(in French with English and Russian summaries)

Abstract: A survey of work done (1955) on the utilization of

furfurol (I) as an intermediate in the production of plastics and pharmaceuticals. The following processes have been developed through the pilotplant stage: (a) the production of pyromicic acid

Card : 1/3

206

HUNGLEY/Chemical Technology. Chemical Products and Their Applications. H
Industrial Organic Synthesis.

Abs Jour: Ref Zhur-Khim., No 8, 1959, 28446.

(II) in yields of 96-98% by the exidation of I with 02 or with air in the presence of Ca(OH)2 and using Ag20 as the catalyst (C); (b) the decarboxylation of II in the vapor phase (C: quinoline) with the separation of furan (III) from CO₂ by adsorption on charcoal; (c) the exidation-decarboxylation of I in the vapor phase (C: exides of heavy netals, particularly Pb); (d) the cleavage of the ring of derivatives of III, particularly of III itself, with H₂O₂ and HCl or with H₂SO₄, leading to the formation of malcic acid (IV) in the first case and of IV and succinic acid, in the second case. Industrial methods for the continuous hydro-

Card : 2/3

HUNGERY/Chemical Technology. Chemical Products and Their Applications. H Industrial Organic Synthesis.

Abs Jour: Ref Zhur-Khin., No 8, 1959, 28446.

genation of I to furfurol alcohol (C: Cu chromite) and of III to tetrahydrofuran have also been developed. -- Ya. Kantor.

Card : 3/3

207.

FODOR, G.

SCIENCE

PERIODICALS. ACTA ZOOLOGICA. Vol. 64, No. 7/8 July/Aug. 1958 AGYAR KEMIAI FILYOTRAT

Fodor, G. Some newer applications of conformation analysis in the chemistry of hydrocarbons. p. 298

Monthly list of mast Buropean Accessions (EEAI) IC, Vol. 8, No. 2, February 1959, Unclass.

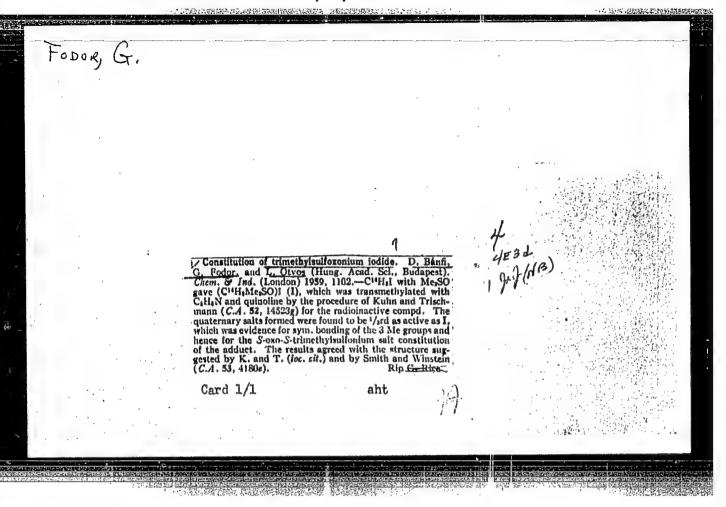
FODOR, G.

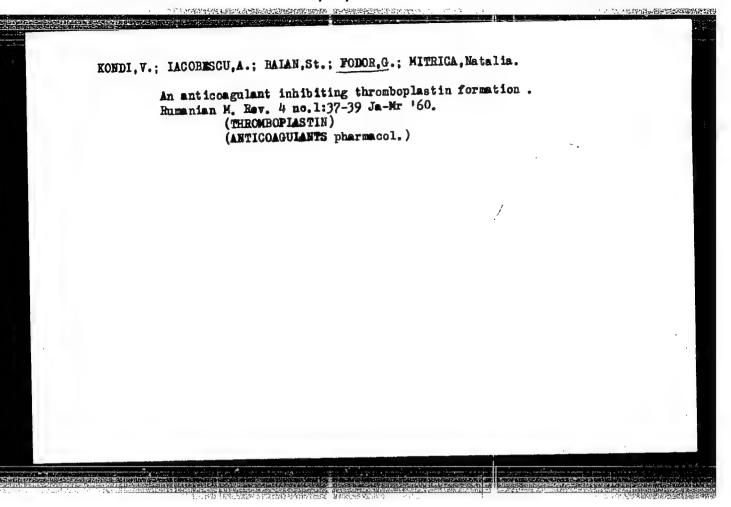
SCIENCE

PERIODICAL: MAGYAR KEMIAI FOLYOIRAT. Vol. 64, no. 7/8, July/Aug. 1958

Fodor, G. Sterochemistry of Prins reaction and its application to ketomes. p. 301.

Monthly list of East European Accessions (EEAI) LC, Vol. 8, No. 2, February 1959, Unclass.





BECK, Mihaly; BITE, Pal; BRUCKNER, Gyozo; CSENTES, Joznef; CSUROS, Zoltan; DEAK, Gyula; ERDEY-GRUZ, Tibor; ERDEY, Iaszlo; FABIAN, Pal; FINALY, Istvan; FODOR, Gabor; FODORNE CSANYI, Piroska; GYORBIRO, Karoly; INZELT, Istvan; KUCSMAN, Arpad; NEUMANN, Erno; PUNGOR, Erno; SCHNEER, Anna; SCHULEK, Elemer; SZABADVARY, Ferenc

Rules for the Hungarian chemical nomenclature and orthography. Kem tud kozl MTA 17 no.1/4:1-292 '62.

1. "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkeszto bizottsagi tagja (for Bruckner, Csuros, Iaszlo Erdey, G.Fodor, and Schulek). 2. "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkesztoje (for Erdey-Gruz). 3. "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" technikai szerkesztoje (for Finaly). 4. Muvelodesugyi Miniszterium (for Csentes). 5. Magyar Tudomanyos Akademia Helyesitasi Bizottsage (for Fabian). 6. Nehezipari Miniszterium (for Neumann).

FODOR, Gabor, akademikus

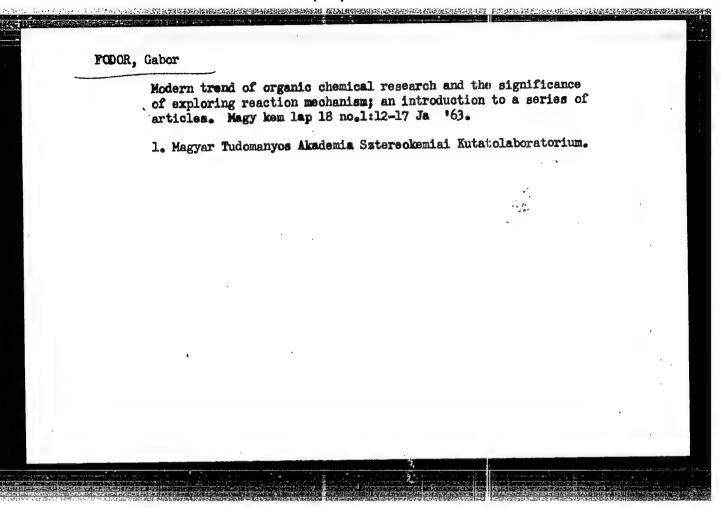
An account of my study trip to England and the German Federal Republic. Kem tud kozl MTA 18 no.2:325-335 '62.

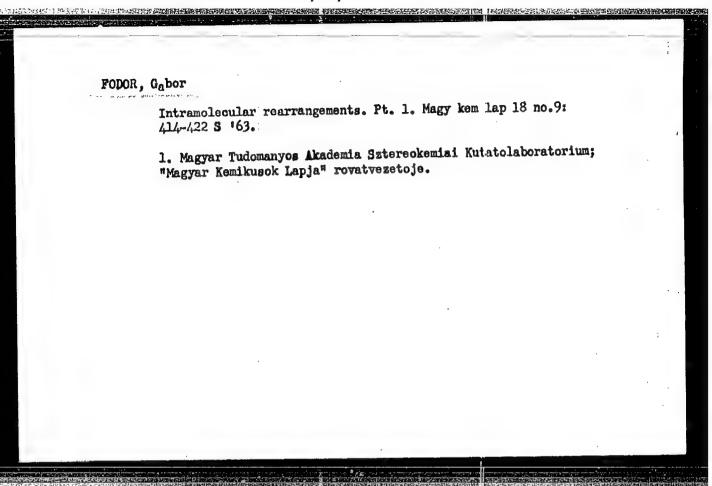
l. Magyar Tudomanyos Akademia Sztereckemiai Kutato Ugoportja, Budapest, es "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkeszto bizottsagi tagja.

FODOR, Gabor, akademikus

An account of the symposium arranged on the \$5th anniversary of the Belgian Chemical Society. Kem tud kozl MTA 18 no.4:605-609 162.

l. Magyar Tudomanyos Akademia Sztereokemiai Kutato Choportja, Budapest, es "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkeszto bizottsagi tagja.





FODOR, Gabor Intramolecular rearrangements, Pt.2. Magy kem lap 18 no.12:593-597 D '63. 1. Magyar Tudomanyos Akademiai Kutato Intezet.

FODOR, Gabor, akademikus; EEKE, Denesne; BITE, Pal, kandidatus; DOBO, Pal;
FARKAS, Lorant, kandidatus; F. VARGA, Eva; IEMPERT, Karoly, kandidatus;
OTVOS, Laszlo, kandidatus; SZANTAY, Csaba, kandidatus; URESCH, Ferenc

An account of the Prague Symposium on Natural Organic Compounds. Kem tud kozl MTA 19 no.1:95-103 '63.

1. Magyar Tudomanyos Akademia Sztereokemiai Kutato Csoportja, Budapest (for Fodor, Beke, Lempert, Otvos, Uresch). 2. Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalya (for Bite, Dobo, Farkas, F. Varga, Szantay). 3. "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkeszto bizottsagi tagja (for Fodor).

FODOR, Gabor, akademikus; MCRACSI, Tivadar; TCMASZ, Jeno

Present state of the chemistry of nucleotides. Kem tud kozl MTA 19 no.2:163-179 '63.

1. Magyar Tudomanyos Akademia Sztereokemiai Kutato Csoportja, Budapest. 2. "A Magyar Tudományos Akademia Kemiai Tudomanyok Osztalyanak Koslemenyei" szerkeszto bizottsagi tagja (for Fodor).

FODOR, Gabor, akademikus

An account of the 1962 Scientific Congress of the Chemical Society in the German Democratic Republic. Kem tud kozl MTA 19 no.3:357-362 163.

1. Magyar Tudomanyos Akademia Sztereokemiai Kutato Csoportja, Budapest; *A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei* szerkeszto bizottsagi tagja.

FODOR, Gabor, akademikus

An account of the First Prague Conference on the Chemistry and Biochemistry of Nucleic Acids, arranged by the scientific academies of the socialist countries. Kem tud kozl MTA 20 no.4:471-472 163.

1. Magyar Tudomanyos Akademia Satereokemiai Kutato Ceoportja, Budapest; A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei zerkeszto bizottsagi tagja.

An account of my study trip to the Soviet Union. Kem tud kosl MTA 19 no.2:239-249 '63. 1. Magyar Tudomanyos Akademia Sstereokemiai Kutato Csoportja, Budapest.

FODOR, Gabor, akademikus

Research in stereochemistry, synthesis and biogenesis of tropane alkaloids conducted since 1955. Pt. 1. Kem tud kozl 20 no.3:336-373 *63.

l. Magyar Tudomanyos Akademia Sztereokemiai Kutato Csoportja, Budapest; "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkeszto bizottsagi tagja.

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Research in the stereochemistry, synthesis and biogenesis of tropane alkaloids since 1955. Pt.2. Kem tud 1951 MYA 20 no.4: 441-467 '63.

1. Magyar Tudomanyos Akademia Setereckemiai Kutato Csoportja, Budapest; "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztanlyanak Közlemenyei" szerkeszto bizottsagi tagja.

ERDEY-GRUZ, Tibor, akademikus; BRUCKNER, Gyozo, akademikus; VAPGHA, larzlo; KORACH, Mor, akademikus; FREUND, Mihaly, akademikus; FODOR, Gabor, akademikus; GERECS, Arpad, akademikus; SCHAY, Geza, akademikus; BITE, Pal, kandidatus; BOGNAR, Rezso, akademikus; FARKAS, Lorand, kandidatus

An account of the work of the Section of Chemical Sciences, Hungarian Academy of Sciences. Kem tud kozl MTA 22 no.2:109-152 '64.

1. Secretary, Section of Chemical Sciences, Hungarian Academy of Sciences, and Editor, "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei", Budapest (for Erdey-Gruz). 2. Editorial board member, "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" (for Bruckner, Korach, Freund, Fodor, Gerecs, Schay and Bognar). 3. Corresponding member, Hungarian Academy of Sciences, and Editorial board member, "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" (for Vargha).

FODOR, Cabor, akademikus

An account of my study trip to Northern Europe. Kem tud kozl MTA
22 no.2:289-291 '64.

1. Research Group of Stereochemistry, Hungarian Academy of
Sciences, Budapest, and Editorial board member, "A Magyar Tudomanyos
Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei".

DYKHOVA, Z.I.; MATYUSHINA, N.A.; MOSKVINA, M.M.; PROKOFTYEVA, G.P.; KHARLAMOV, V.T.; CHIRKOV, Ye.P.; FODOR, G.; FILLIF, I.

[Radioactive isotopes and labeled compounds; a catalog] Radioaktivnye isotopy i mechenye soedineniia; katalog. Moskva, Atomizdat, 1964. 341 p. (MIRA 18:1)

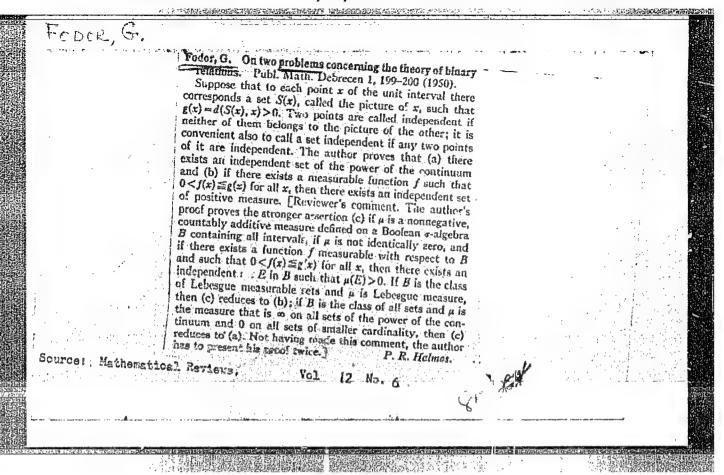
1. Sovet ekonomicheskoy vzaimopomoshchi. Postoyannaya kemissiya po ispolizovaniyu energii v mirnykh tselyakh.

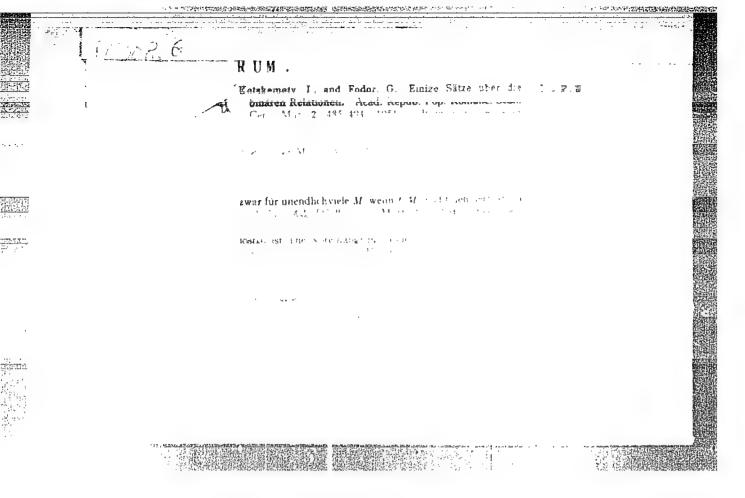
GALATEANU. I.: FODOR, G.; CHIOTAN, C.; CRISTU, M.

Obtaining 59Fe without a bearer. Studii cerc chim 13 no.10:643-652 0 '64.

1. Institute of Atomic Physics of the Rumanian Academy, Bucharest, P.O. Box 35.

APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000413330013-3"





Fodor, G. Proof of a conjecture of P. Erdős, Acta Sci.
Math. Szeged 14, 219-227 (1952).
Soit E un ensemble non dénombrable de puissance m et n
un nombre cardinal donné, tel que Na n. Si RCEXE
est une relation binaire entre éléments de E telle que
card R(x) < n, R est coloriable nvec moins deut couleurs c'està-dire, il existe une relation d'équivalence UCEXE telle que
RNUCA et card (E/U) n. Ce théorème, qui avait cté
conjecturé par P. Erdős [Proc. Amer. Math. Soc. 1, 127-141]
(1950), pp. 133-137; ces Rev. 12, 14] donne immédiatement
une démonstration de la conjecture de Ruziewicz (c'est-àdire, il existe un sous ensemble XCE tel que RN(XX)CA
et card X = n) dans le cas où m ne peut être décomposé en
une somme de n ou d'un nombre moindre de nombres
cardinaux dont chacun est plus petit que m. J. Riguet.

Mathomatical Reviews Vol. 15 No. 2
Feb. 1954
Analysis

Following two propositions are equivalent: (1) $2^{N_a} = N_{a+1}$. (2) Let $|E| = 2^{N_a}$, and denote by B the class of all subsets of E of power 2. Then there exists a mapping, T, of B into E such that (a) if $r = |x_a| > 1$, $r = |x_a| > 1$, then then E is equal to the union of the sets $r \in B$ for which $T(r) \in E_1$.

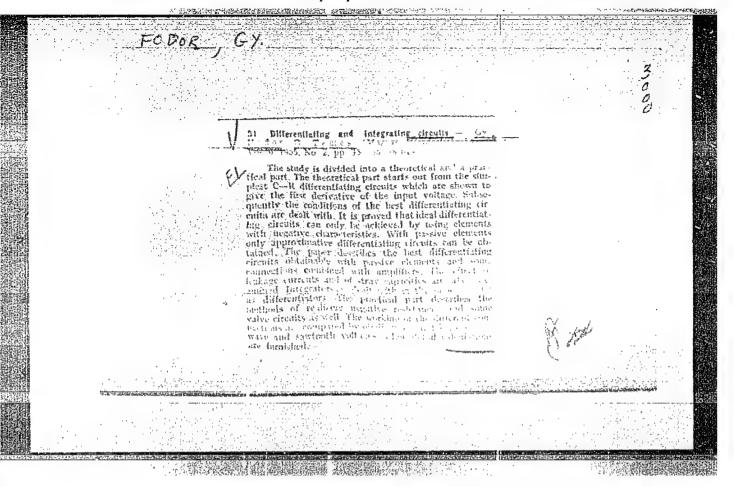
Foldor, G. An assertion which is equivalent to the generalized continuum hypothesis. Acta Sci. Math. Szeged 15, 17-18 (1953).

For every ordinal number a, the following two propositions are equivalent: (1) $2^{N_a} = N_{a+1}$. (2) Let $|E| = 2^{N_a}$, and denote by B the class of all subsets of E of power 2. Then there exists a mapping, T, of B into E such that (a) if $r = |x_a| > 1$. F, Bagemihl.

FODOR, G. - Koslemenyei - Vol. 5, no. 1, 1955.

Problem of the set theory. p. 57.

SO: Monthly list of East European Accessions, (EEAL), LC, Vol. 4, No. 9, Sept. 1955 Uncl.



FODOR, C.

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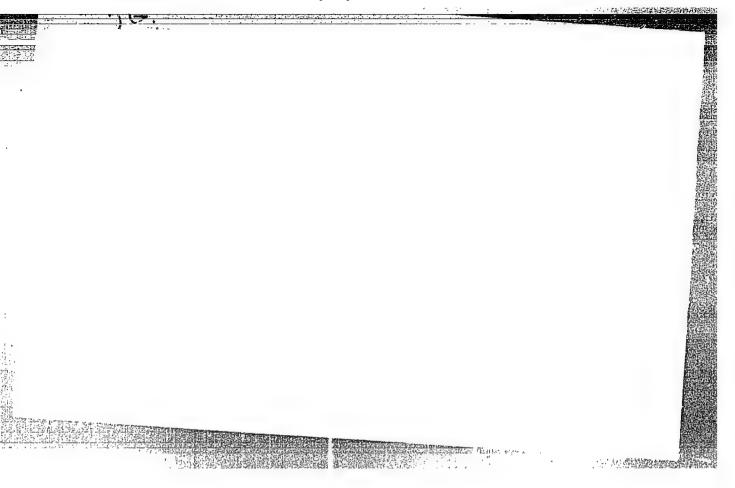
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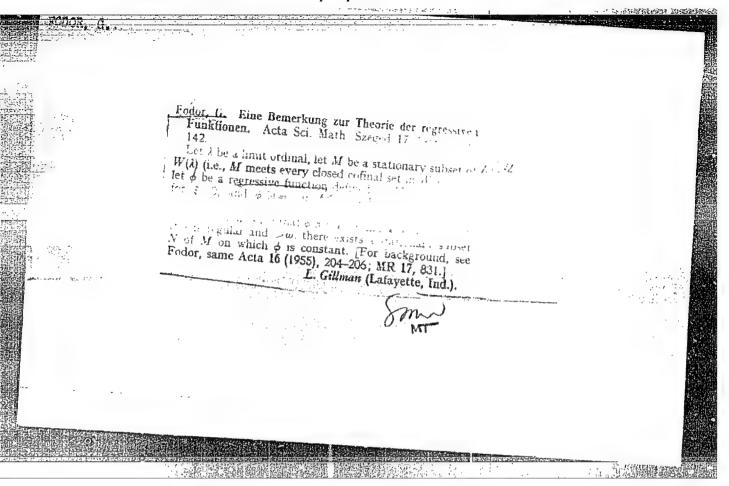
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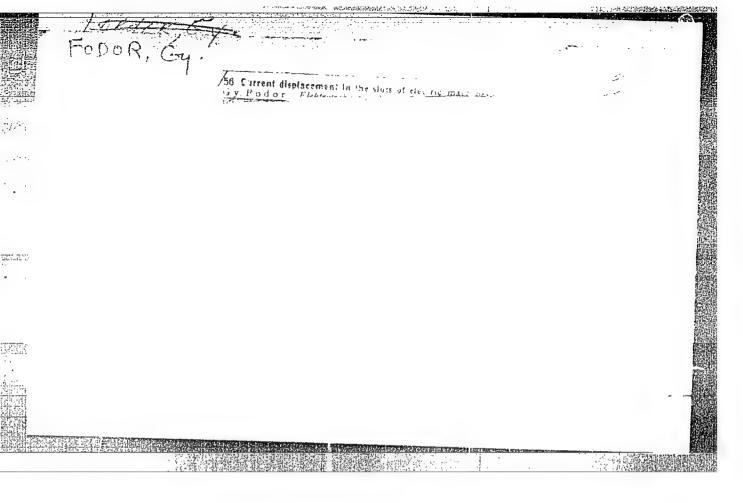
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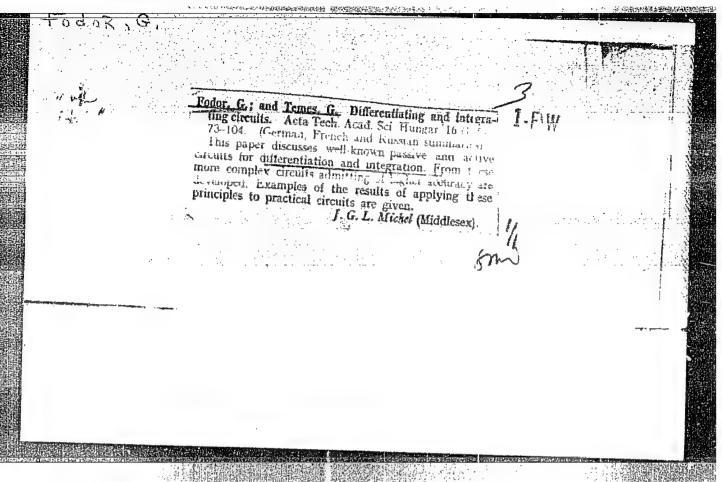
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Regulation of reactors. I. (To be contd.)

P. 582. (ENERGIA ES ATCMTECHNIKA.) (Budapest, Hungary) Vol. 10, No. 11/12, Nov./Dec. 1957

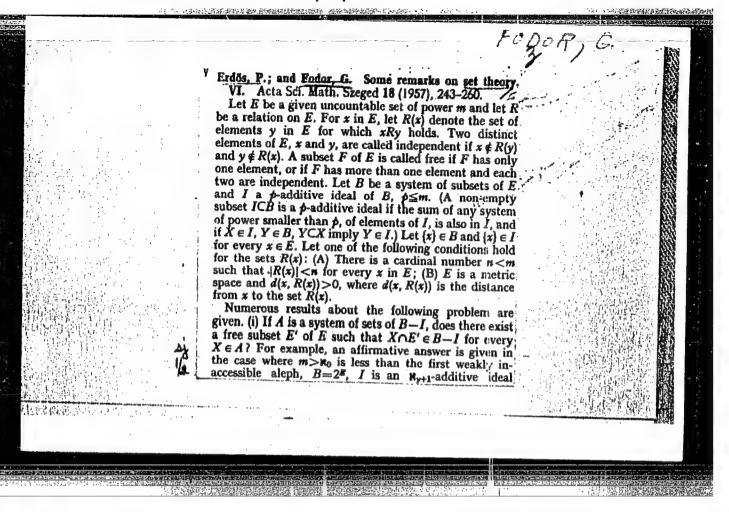
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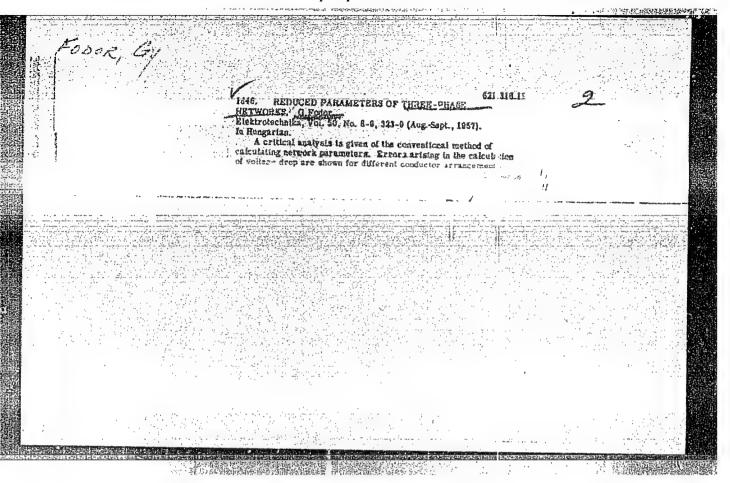


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C-8

Abs Jour : Ref Zhur - Fizike, No 4, 1959, No 7767

Author : Fodor Gyorgy

Inst : -

Title : Control of Reactors, Part II.

Ori $_{\mathbb{S}}$ Pub : Energia es atomtechn., 1958, 11, No 1-2, 1-8

Abstract : Survey article on the control of reactors. The following

problems are considered: self-regulation of the reactor, control rods, the reactor-control loop, the transfer function, programed regulation, and starting and stopping of

the reactor. -- V.I. Lend'yel

Card : 1/1

18

HUNG:RY/Nuclear Physics - Nuclear Power and Technology

c-8

Abs Jour : Ref Zhur - Fizika, No 4, 1959, No 7736

Author : Fodor Gyorgy

Inst Title

: -.

: Dictionary in Nuclear Engineering

Orig Pub: Energie es atomtechn., 1958, 11, No 1-2, 39-40

Abstract : An explanation is given of many terms pertaining to reactors.

Card : 1/1

16

HUNGARY/Nuclear Physics - General

C-1

Abs Jour: Ref Zhur - Fizika, No 3, 1959, No 4933

Author

: Fodor Gyorgy

Inst

Title

: Systems of Units in Atomic Engineering

Orig Pub: Energia es Atomtechnika, 1958, 11, No 3, 138-142

Abstract: The author analyzes the MKS, CGS and the practical atomic system, the technical system of absolute units, and the so-called modified absolute system. A table of conversion of various quantities from one system to another is given. The author draws the following conclusion from his premises: for physical problems it is best to use the system MkJ, CGS, the practical atomic, and possible also the modified atomic; for technical problems it is best to use the MS or the

technical system of units. -- V.I. Lend'yel

: 1/1 Card

HUNGARY/Nuclear Physics - Nuclear Technology and Power Engineering C-8

Abs Jour : Ref Zhur - Fizika, No 4, 1959, No 5273

Author : Fodor Gyorgy

Inst T

Title : Dictionary of Nuclear Engineering [sic!]

Orig Pub: Energia es Atomtechnika, 1958, 11, No 3, 151-152

Abstract : Expressions are given for the critical demensions of the

sphere, cylinder, and tube as functions of the reactor para-

meter. Tables are given for the physical constants of

heavy water and of its chemical properties.

Card : 1/1

FODOR, G.

HUNGARY/Nuclear Physics - Penetration of Charged and Neutral Particles Through Matter

c-6

Abs Jour : Ref Khur- Fizika, No 5, 1959, No 10186

: Fodor Gyorgy Author

Inst

: Measurement of the Diffusion Length in Dodies Having a Title

Shape of a Prism, Cylinder, or Sphere

Orig Pub : Energia es Atomtechn., 1958, 11, No 4-5, 294-302

Abstract : No abstract

: 1/1 Card

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Temperature factor, p.650

EMERGIA ES ATONTECHNIKA: (Emergiagazdalkodasi Tudomanyos Egyesulet) Budapost, Hungary Vol. 11, no.9/10, Sept./Oct. 1958

Monthly List of East European Accessions (SEAI) LC., Vol. 8, no.7, July 1959 Uncl.

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Nuclear technical encyclopedia. p673.

ENERGIA ES ATOMTECHNIKA. (Energiagazdalkodasi Tudomanyos Egyesulet)
Budapest, Hungary
Vol. 11, no.11/12, Nov./Dec. 1958

Monthly List of East European Accessions (EEAI) LC., Vol. 8, no.7, July 1959

FODOR, G. (Budapest, XI., Budafoki ut 6-8)

The interpretation of characteristics of fundamental equations of the electromagnetic field. Periodica polytechn electr 3 no.3: 197-215 *59. (EEAI 10:1)

1. Budapest Polytechnical University Institute for Theoretical Electricity.

(Electromagnetic fields) (Equations)

Alternating current impodance of sheeted conductors. 1.1.5.

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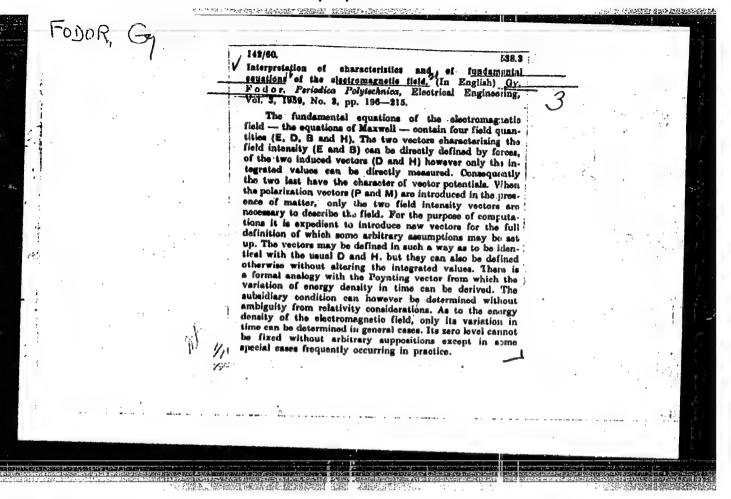
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Fodor, Gy.

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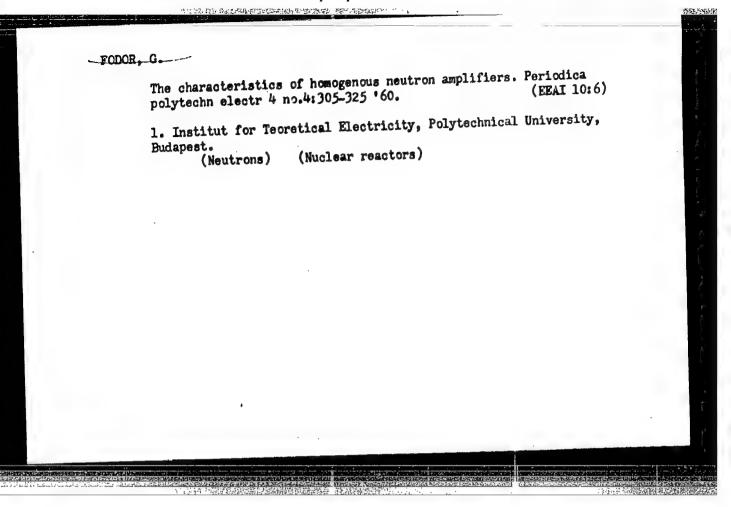
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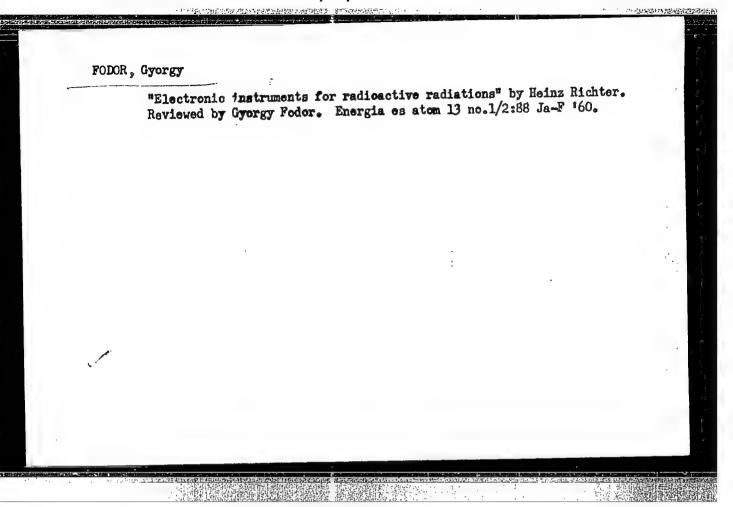
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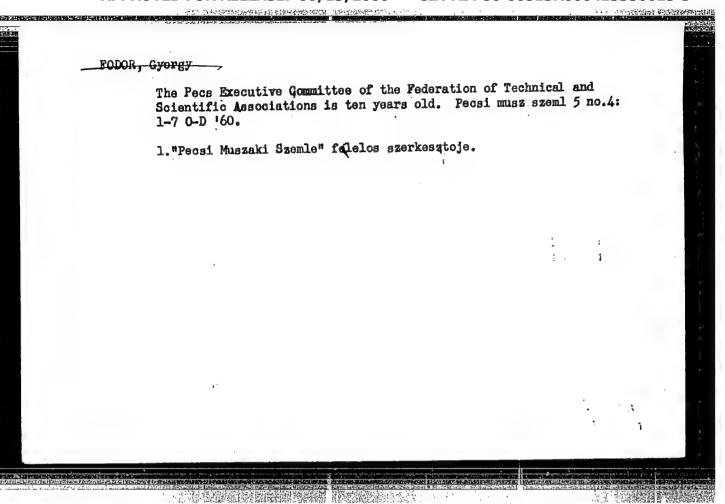
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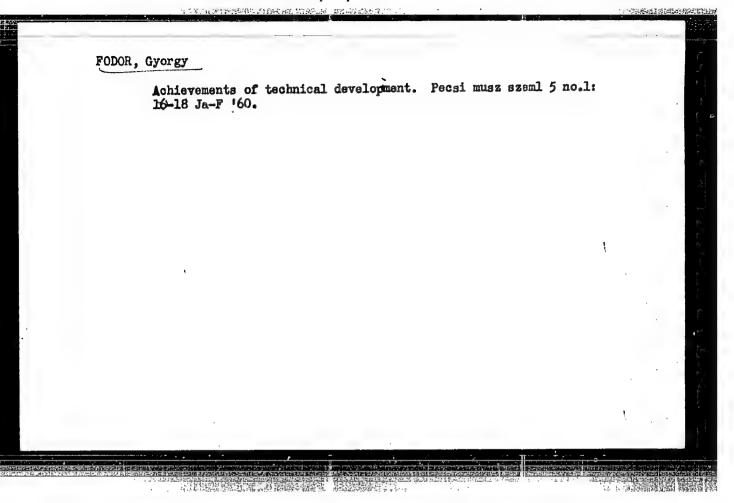
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1. Department for Theoretical Electricity, Polytechnical University, Budapest. Presented by Prof. Dr. F. Csaki.

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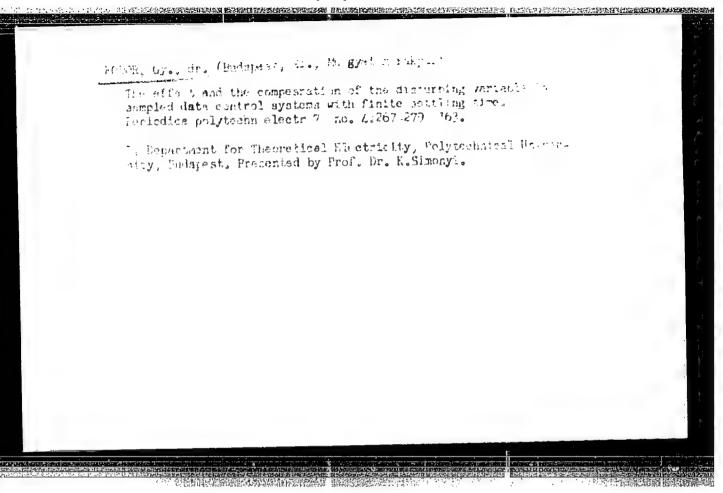
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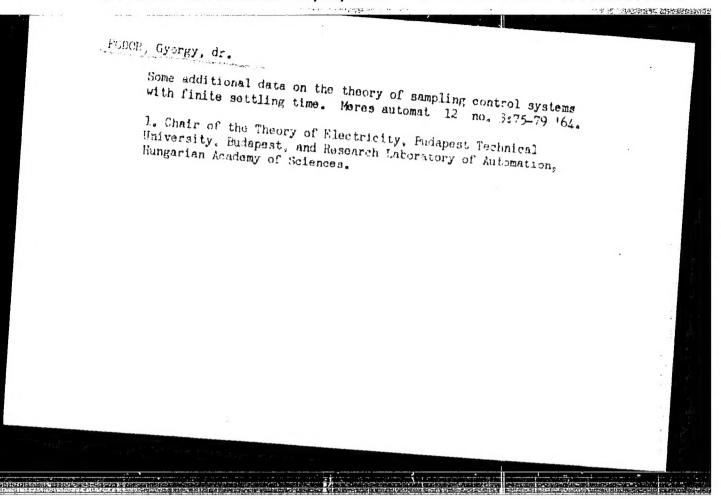
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